

Thermal behavior of pseudovaterite NdBO₃E.N. Sokolova¹, R.S. Bubnova², S.K. Filatov¹, M.G. Krzhizhanovskaya¹¹*Department of Crystallography, St. Petersburg State University, Russia*²*Institute of the Silicate Chemistry of the Russian Academy of Sciences, St. Petersburg, Russia*

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There are known pseudovaterite, aragonite and H-modifications of NdBO₃ [1-3]. Here we present thermal behavior of pseudovaterite NdBO₃ including its thermal expansion.

Polycrystalline sample of pseudovaterite NdBO₃ was prepared through hydrothermal method. Starting materials were high purity Nd₂O₃ and H₃BO₃. Reagents were mixed in stoichiometric proportions and finely grounded, and then the charge was put to autoclave with 40 ml of NaOH solution (0.2 M) at the temperature 493 K for 2 weeks. At the result a metastable hexagonal phase of NdBO₃ pseudovaterite type first discovered in [1] has formed according to X-ray powder diffraction data. Thermal behavior of the sample has been studied in air from 293 to 1163 K by means of high-temperature X-ray powder diffraction data collected using a Stoe Stadi P X-ray diffractometer (CuK α radiation) with a high-temperature camera (Buehler GmbH).

The HTXRD patterns of pseudovaterite remain without any changes up to 973 K. On further heating peaks of the aragonite modification appear. The thermal expansion is determined in linear approximation in temperature range of 293-943 K before phase transition beginning. Thermal expansion of pseudovaterite NdBO₃ is close to isotropic one: $\alpha_a = 11.5$, $\alpha_c = 10$, $\alpha_V = 33$, $\langle\alpha\rangle = \alpha_V/3 = 11 \times 10^{-6} \text{ C}^{-1}$. Pseudovaterite NdBO₃ shows average linear thermal expansion close to that of other M³⁺ borates for example it is about $12.7 \times 10^{-6} \text{ C}^{-1}$ for Bi³⁺-borates [4].

The crystal structure of this phase has not been yet determined up to now. There are two structural models for the hexagonal phase of MBO₃ where $M = Y$ and RE (Sm-Lu) [1, 5]. Both models are built up from alternative layers of and BO- and REO-polyhedra along c axis: one present crystal structure with isolated BO₃ triangles planes of those disposed parallel to c -axis [4], another - with isolated tetrahedra [3]. REO-polyhedra sharing corners form framework in both models. By high-temperature borate crystal chemistry principles [4, 6] if crystal structure is based on isolated BO₃ triangles the most thermal expansion should be expected in directions to perpendicular to a plane of triangles and minimal to parallel the plane of BO₃; in the case of crystal structure with isolated BO₄ tetrahedra thermal expansion is expected close to isotropic. Hence thermal expansion pseudovaterite NdBO₃ is good agreement with both crystal structures: pseudovaterite minimally expands along c axis and slight more intensive along a axis and such behavior of expansion could be caused by disposition of the BO₃ triangles parallel to c axis while the expansion is close to as it should be expected in case of existence of BO₄ tetrahedra.

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